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Desai Hiralben V
 Department of Biotechnology
 and Biochemistry, Junagadh
 Agricultural University,
 Junagadh, Gujarat, India

Mandavia MK
 Department of Biotechnology
 and Biochemistry, Junagadh
 Agricultural University,
 Junagadh, Gujarat, India

Nidhi Radadiya
 Department of Biotechnology
 and Biochemistry, Junagadh
 Agricultural University,
 Junagadh, Gujarat, India

Jadav JK
 Department of Biotechnology
 and Biochemistry, Junagadh
 Agricultural University,
 Junagadh, Gujarat, India

Golakiya BA
 Department of Biotechnology
 and Biochemistry, Junagadh
 Agricultural University,
 Junagadh, Gujarat, India

Correspondence
Desai Hiralben V
 Department of Biotechnology
 and Biochemistry, Junagadh
 Agricultural University,
 Junagadh, Gujarat, India

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Metabolite profiling of mango cv. Kesar leaf using gas chromatography-mass spectrometry

Desai Hiralben V, Mandavia MK, Nidhi Radadiya, Jadav JK and Golakiya BA

Abstract

Mango Cv. Kesar is most popular variety of Gujarat. The leaf metabolite profiling of this cultivar carried out using highly used platform Gas chromatography-Mass spectrometry (GC-MS). The leaf sample metabolites were extracted and derivatized for metabolite profiling and identified on the bases of Retention time (min) and mass to charge ratio (m/z). These metabolites were classified in organic Acid (41% of total metabolites), Sugars (32%), other class (18%) and smallest class sugar alcohol (9%). Metabolite profiling of Mango Cv. Kesar leaf revealed that, the potential metabolites have their significance function in plant metabolism, plant protection against biotic and abiotic stress and also important for human health.

Keywords: GC-MS, mango Cv. Kesar, metabolites, organic acids, sugars, sugar alcohols

Introduction

The metabolic networks in higher plants are highly complex and multiplex biochemical steps. Metabolomics approaches used to assess the natural variation in metabolite content in individual plants, with great potential for the improvement of the compositional quality of crops. Metabolomics analysis is typically performed by nuclear magnetic resonance spectroscopy (NMR) and mass spectrometry (MS), including gas chromatography GC-MS, liquid chromatography LC-MS, capillary electrophoresis CE-MS among these platforms GC-MS is one of the mostly widely used analytical methods for metabolomics. Metabolite profiling is a fast growing technology and is useful for phenotyping and diagnostic analyses of plants. It is also rapidly becoming a key tool in functional annotation of genes and in the comprehensive understanding of the cellular response to biological conditions. (Gullberg *et al.*, 2004; Schauer and Fernie, 2006) [8, 18]. Mango is the national fruit of India, known as the 'King of Fruits'. It is one of the most important and popular Asian fruits. Mango is popular due to its excellent flavour, delicious taste, delicate fragrance, attractive colour and nutritive value which make at rank among the best fruits of world. It is believed that mangoes originated in Northeast India. Mango (*Mangifera indica* L.) universally considered to be one of the finest fruits, and is an important crop in tropical and subtropical areas of the world. Major mango growing countries are India, China, Thailand, Pakistan, Australia, Indonesia, Bangladesh, Philippines, Nigeria, Myanmar and Egypt. There are about 1500 varieties of mango in the world from which about 1200 are found in India (Krishnan *et al.*, 2009) [10]. The most popular cultivar grown around Gujarat state is Mango cv. Kesar. Kesar is characterized by its golden color with green over tones, and its unique flavour. The fruit is slightly smaller compared to the Alphonso variety. The fruits are medium to large sized (250-325 g per fruit), oblong in shape with an attractive light apricot-yellow color. The taste is very good and sugar/ acid blend is excellent. The cultivar is free from spongy tissue disorder and malformation. Tree bear excellent quality fruits with saffron coloured pulp when ripe and delicious. Excellent for table purpose fruits, medium sized with fiber-less stone. The "Kesar" fruit has 18 to 22 percent T.S.S., 0.25 to 0.29 percent acidity and 10.5 to 12.0 percent total sugars with storability of 15 to 20 days (Singh, 1960; Chovatia *et al.*, 1995) [20, 6]. Ample information on Mango cv. Kesar leaf metabolite profiling available. Here, we characterized the contribution of metabolite profiling of Mango Cv. Kesar leaf and this information could be valuable to characterize certain aspects of genetic traits, which in turn will be useful for breeding programs and diagnostic/treatments aspects in future.

Materials and Methods

Mango cv. Kesar leaf sample was procured from Fruit research station, Sakkarbaug Farm, Junagadh Agricultural University, Junagadh, Gujarat, India. The experiment was carried out at department of Biotechnology, Junagadh agricultural University, Junagadh, Gujarat.

Extraction and derivatization of metabolites

Metabolite profiling study was performed using GC–MS. Metabolites were extracted as described by Liseć *et al.* (2006) [12] with minor modifications.

1. Leaf tissues (150 mg) were homogenized with pre-chilled mortar-pestle in 3 ml of 100% HPLC grade methanol (precooled at -20°C).
2. The mixture was shaken for 10 min at 70°C in a water-bath at 950 rpm and centrifuged for 10 min at 11,000 g. The supernatant was transferred to a Schott GL14 glass vial and 1.5 ml of chloroform (-20°C) was added.
3. After that 3.0 ml of dH_2O (4°C) was added and vortexed for 10s.
4. Again, the mixture was centrifuged for 15 min at 2200 g and the upper phase (polar) and lower phase (nonpolar) phase were transferred into a separate test tube.
5. Both polar and nonpolar phase were dried in a nitrogen stream. Extracted metabolites were derivatized as described by Sanimah *et al.* (2013) [17] with minor modifications.
6. The dried extracts were re-dissolved in 50 μl of pyridine and sonicated for 10 min.
7. Then, 100 μl of methoxyamine HCL (20 mg ml⁻¹ in pyridine) was added and vortexed for 30s.
8. The mixtures were then sonicated again for 5 min and incubated with constant agitation for 90 min at 37°C .

9. The trimethylsilylation (TMS) step was performed by adding 250 μl *N*-Methyl-*N*-(trimethylsilyl) trifluoroacetamide (MSTFA) to the extracts and vortexed for 30s.

10. Mixtures were incubated for 1 h at 37°C for derivatization.

GC-MS analysis

For GC–MS analysis, 1 μl of derivatized extract was injected into a DB-17MS capillary (30 m \times 0.25 mm). The inlet temperature was set at 280°C . After a solvent delay for 5 min, initial GC oven temperature was set at 100°C ; after injection for 1 min, the GC oven temperature was raised to 290°C .

The injection temperature was set to 280°C and ion source temperature was 230°C . Helium was used as the carrier gas with a constant flow rate set at 1 ml/min. The measurement was performed with electron impact ionization (70 eV) in the full scan mode (*m/z* from 50 to 700). Metabolites were putatively identified by matching their mass spectra to spectra in NIST 14 library (National Institute of Standards and Technology, Gaithersburg, MD, USA). Pre-processing of total ion chromatograms (TIC) such as baseline correction, alignment, peak picking, and integration were performed using the ACD/Spec Manager v.12.00 (Advanced Chemistry Development, Inc., ACD/Labs, Toronto, Canada). CSV comma delimited files were created for data analysis.

Results and Discussions

Total 85 different metabolites were identified in leaf of Mango Cv. Kesar by GC-MS. These metabolites were identified on the bases of their Retention Time (min) and *m/z* (mass to charge) ratio. The GCMS chromatogram obtained from leaf of Mango Cv. Kesar is presented in Fig. 1.

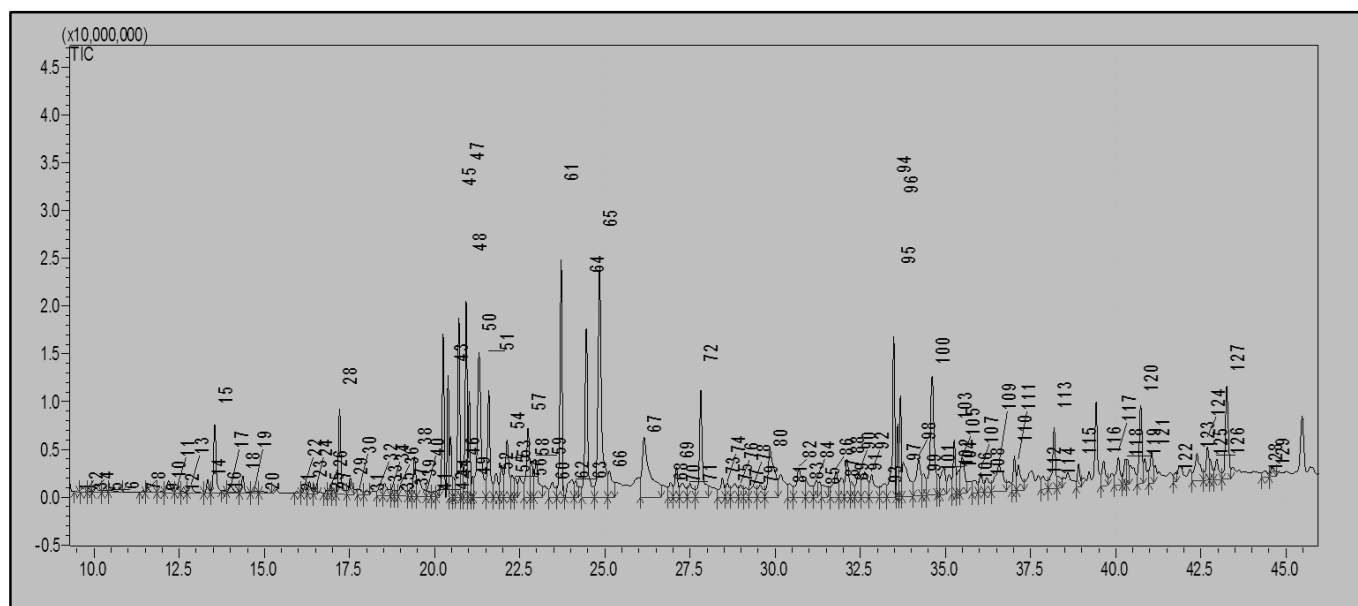


Fig 1: GCMS chromatogram of Mango Cv. Kesar leaf.

Functional classification and significance

The largest class was organic acid followed by, sugar, other classes and sugar alcohol (Fig. 2, Table 1). The largest class “Organic acid” comprised 41%. Organic acid involved at the cellular level for several biochemical pathways at the whole plant level in modulating adaptation to the environment (Lopez-Bucio *et al.*, 2000) [13]. This class includes few of the well known metabolites putatively known for their medicinal

as well as economic values further strengthening the medicinal and economic properties of Kesar leaf. Among these organic acids, 1-Cyclohexene-1-carboxylic acid (RT 21.018) also known as Shikimic acid, a natural organic compound is generally utilized as a precursor for industrial synthesis of the antiviral Oseltamivir drug, chemotherapy of cancerous diseases, used as herbicides and antibacterial agents and for benzene-free production of phenol (Bochkov *et al.*,

2012). Threonic acid (2, 3, 4-Trihydroxybutyric acid, RT 14.367) is a sugar acid derived from threose, L-isomer is a metabolite of ascorbic acid (England and Seifter., 1986) [7]. One study suggested that because L-threonate inhibits DKK1 expression *in vitro*, it may have potential in treatment of androgenic alopecia (Kwack *et al.*, 2010) [11]. Other important members of this group were 2-Butenedioic acid, 2-Pentenedioic acid, Benzoic acid, ribonic acids, malic acid, L-Ascorbic acid etc. have their particular role in metabolism, growth and development of plant.

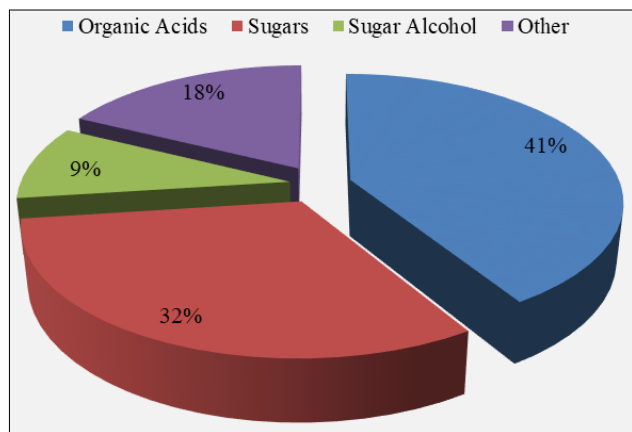


Fig 2: Functional classification of metabolites. All the metabolites identified by GC-MS analysis was classified into four groups as represented in the pie chart

Second largest class of metabolites was the “Sugars”, which comprised 32% of total metabolites. Sugars role as providers of carbon and energy and fulfills signaling role in coordination with hormonal signaling pathways (Rolland *et al.*, 2006) [16] controlling various plant physiological processes, including innate immunity (Bolouri Moghaddam and Van den Ende, 2012) [4]. Glucose, fructose, Galactose, maltose, ribose, turanose, melibiose and their different variants were present in leaf of Mngo Cv. Kesar. These are natural sugar derivatives and their different isoforms or variants were present at different retention time which given in Table 1.

The third largest class of metabolites was the “Other classes” which covers 18% of total metabolites, which includes various intermediate products of different metabolic pathways and their byproducts. Few of the metabolites and/or their isoforms eluted at different retention time in given in Table 1. This class includes 1-Monooleoylglycerol trimethylsilyl ether, Furanone, nucleoside uridin and psudouridin and alcohol derivatives. Uridine, a nucleoside was omnipresent across the wild type and transgenic lines. It contains a uracil attached to a ribose ring (known as a ribofuranose) via a β -N1-glycosidic bond. Uridine plays a major role in linking the glycolysis pathway with that of galactose and the conversion of

galactose to glucose. Uridin containing drug Souvenaid® has demonstrated cognitive benefits in patients with mild Alzheimer's disease (Ritchie *et al.*, 2014) [15]. Monobe *et al.* (2003) studies suggested metabolite Pseudo uridine may offer protection from radiation. Furanones are also important as naturally occurring flavor compounds. They are found in many fruits including strawberry, pineapple, raspberry, grapes, tomato, kiwi and mango and many contribute caramel-like flavor notes of these fruits (Schwab and Roscher 1997) [19]. Ripe mango fruits contain high amounts of furaneol (4-hydroxy-2, 5-dimethyl-3(2H)-furanone) and its methyl ether, mesifuran (2, 5-dimethyl-4-methoxy-3(2H)-furanone). The fruits of cultivar Alphonso contained higher amounts of these two compounds than those found in any other cultivar (Pandit *et al.* 2009) [14]. Squalene (2, 6, 10, 14, 18, 22-Tetracosahexaene, 2, 6, 10, 15, 19, 23-hexamethyl-, (all-E)-, RT 40.733) has some beneficial properties when consumed either from food or as a supplement. These include potential anticancer effects, protective and moisturizing role in the skin, antioxidant activity and it may even improve the immune response to vaccines.

Furthermore, the smallest class of Mango Cv. Kesar leaf metabolome encompasses sugar alcohols. Sugar alcohols are reduced forms of aldose or ketose monosaccharide (Brimacombe and Webber, 1972) [5]. Mannitol, glucitol, myo-inositol and arabitol were the sugar alcohols found to be present in the leaf metabolome. Sugar alcohols are a type of carbohydrates called “polyols” and are used as subdued calorie sweetener, and often in combination with high intensity artificial sweeteners to counter the low sweetness. These sugar substitutes offer fairly low calories than table sugar (sucrose), primarily due to its low absorption and may even have a small laxative effect. Besides this function sugar alcohol also involved in osmoprotective mechanism of plant against Abiotic stress such as cold, drought, salt stress. The transgenic tobacco containing mannitol 1-phosphate dehydrogenase (mtID) from *Escherichia coli* resulted in mannitol production and a salinity-tolerant phenotype (Tarczynski *et al.*, 1993; Thomas *et al.*, 1995) [21, 22]. The cyclitols such as myo-inositol have shown better stress protection. The myo-inositol is synthesized by myo-inositol 1-phosphate synthase and is induced by salinity (Ishitani *et al.*, 1996) [9]. The transgenic *Arabidopsis* plants overexpressing *G.max* IMT have displayed improved tolerance to dehydration stress treatment and high salinity stress treatment (Ahn *et al.*, 2011) [1]. The transgenic *A. thaliana* plants expressing halophyte *Mesembryanthemum crystallinum* myo-inositol-O-methyltransferase (Imt1) in response to cold stress have shown elevated cold tolerance. In addition, accumulation of sugar alcohols like mannitol or sorbitol has been linked to stress tolerance (Arbona *et al.* 2013) [2].

Table 1: List of Metabolites in leaf of Mango Cv. Kesar

| Organic Acids | | |
|---|--------|-----------|
| Name | RT | m/z ratio |
| 1-Cyclohexene-1-carboxylic acid, 3, 4, 5-tris [(Trimethylsilyl)oxy]-, trimethylsilyl ester, [3R-(3.alpha., 4.alpha., 5.beta.)]- | 21.018 | 204.15 |
| 2, 3, 4-Trihydroxybutyric acid tetrakis(trimethylsilyl) deriv. | 14.367 | 73.1 |
| 2-Butenedioic acid (E)-, bis(trimethylsilyl) ester | 9.749 | 245.15 |
| 2-Butenedioic acid (Z)-, bis(trimethylsilyl) ester | 10.499 | 147.15 |
| 2-Pentenedioic acid, 2-[(trimethylsilyl)oxy]-, bis(trimethylsilyl) ester | 22.98 | 73.1 |
| 2-Trimethylsilyloxyheptanoic acid, trimethylsilyl ester | 9.472 | 173.25 |
| 3-Ethyl-3-hydroxyglutaric acid triTMS | 22.743 | 73.1 |

| | | |
|--|--------|--------|
| 3-Hydroxy-3-(4'-hydroxy-3'-methoxyphenyl)propionic acid, tri-TMS | 37.868 | 297.15 |
| Benzoic acid, 3,4,5-tris(trimethylsiloxy)-, trimethylsilyl ester | 24.844 | 73.05 |
| Benzoic acid, 3,4,5-tris(trimethylsilyloxy)-, propyl ester | 29.352 | 73.05 |
| Benzoic acid, 3-methyl-2-trimethylsilyloxy-, trimethylsilyl ester | 24.455 | 281.15 |
| Benzoic acid, 4-[(trimethylsilyloxy)-, trimethylsilyl ester | 18.41 | 73.1 |
| Butanedioic acid, bis(trimethylsilyl) ester | 9.877 | 147.15 |
| D-Gluconic acid, 2,3,4,5,6-pentakis-O-(trimethylsilyl)-, trimethylsilyl ester | 19.009 | 73.1 |
| D-Glucuronic acid, 2,3,4,5-tetrakis-O-(trimethylsilyl)-, trimethylsilyl ester | 31.212 | 73.05 |
| Gluconic acid, gamma.-lactone, 2-methoximine, tri(trimethylsilyl)- | 24.03 | 73.1 |
| Gluconic acid, gamma.-lactone, 5-methoximine, tri(trimethylsilyl)- | 25.13 | 73.1 |
| Gluconic acid, 2-methoxime, tetra(trimethylsilyl)-, trimethylsilyl ester | 17.784 | 73.1 |
| Gluconic acid, 2-methoximine, tetra(trimethylsilyl)-, trimethylsilyl ester | 28.816 | 73.1 |
| Gulonic acid, 2,3,5,6-tetrakis-O-(trimethylsilyl)-, lactone | 19.782 | 73.1 |
| Hexacosanoic acid, 2-[(trimethylsilyloxy)-, trimethylsilyl ester | 40.08 | 439.5 |
| Hexadecanoic acid, 2,3-bis[(trimethylsilyloxy)propyl ester | 35.384 | 73.05 |
| Hexadecanoic acid, trimethylsilyl ester | 26.151 | 73.05 |
| Idonic acid, 2,3,5,6-tetrakis-O-(trimethylsilyl)-, lactone | 22.36 | 73.1 |
| L-Ascorbic acid, 2,3,5,6-tetrakis-O-(trimethylsilyl)- | 24.14 | 73.1 |
| Malic acid, tris(trimethylsilyl) ester | 13.543 | 73.1 |
| Malonic acid, bis(2-trimethylsilylethyl ester | 21.115 | 73.1 |
| Mannonic acid, 2,3,4,6-tetrakis-O-(trimethylsilyl)-, lactone | 22.841 | 73.1 |
| Nonanoic acid, trimethylsilyl ester | 10.061 | 73.1 |
| Octanedioic acid, bis(trimethylsilyl) ester | 19.989 | 73.1 |
| Propanoic acid, 2,3-bis[(trimethylsilyloxy)-, trimethylsilyl ester | 18 | 73.1 |
| Propanoic acid, 2-methyl-2,3-bis[(trimethylsilyloxy)-, trimethylsilyl ester | 11.989 | 73.1 |
| Ribonic acid, 2,3,4,5-tetrakis-O-(trimethylsilyl)-, trimethylsilyl ester | 18.528 | 73.1 |
| Talonic acid, 2,3,5,6-tetrakis-O-(trimethylsilyl)-, lactone | 38.911 | 73.1 |
| trans-9-Octadecenoic acid, trimethylsilyl ester | 29.85 | 73.05 |
| Sugar | | |
| .alpha.-D-Mannopyranose, 1,2,3,4,6-pentakis-O-(trimethylsilyl)- | 32.832 | 204.1 |
| .beta.-D-Galactofuranose, 1,2,3,5,6-pentakis-O-(trimethylsilyl)- | 22.277 | 217.15 |
| .beta.-D-Glucopyranose, 1,2,3,4,6-pentakis-O-(trimethylsilyl)- | 31.783 | 204.1 |
| 2-Deoxy ribose per-TMS II | 15.956 | 73.1 |
| 4-Ketoglucose, bis(O-methyloxime), tetrakis(trimethylsilyl) | 12.446 | 73.1 |
| 4-Ketoglucose, methoxy, silyl | 21.588 | 73.1 |
| Arabinofuranose, 1,2,3,5-tetrakis-O-(trimethylsilyl)- | 19.534 | 217.15 |
| Arabino-Hexos-2-ulose, 3,4,5,6-tetrakis-O-(trimethylsilyl)-, bis(dimethyl acetal) | 12.581 | 73.1 |
| d-Arabinose, tetrakis(trimethylsilyl)- | 19.295 | 73.1 |
| D-Fructose, 1,3,4,5,6-pentakis-O-(trimethylsilyl)-, O-methyloxime | 20.709 | 73.1 |
| D-Fructose, 6-O-[2,3,4,6-tetrakis-O-(trimethylsilyl)-.alpha.-D-glucopyranosyl]-1,3,4,5-tetrakis-O-(trimethylsilyl)- | 44.376 | 73.1 |
| D-Galactose, 6-deoxy-2,3,4,5-tetrakis-O-(trimethylsilyl)-, O-methyloxime | 17.011 | 117.15 |
| D-Mannopyranose, 1,2,3,4,6-pentakis-O-(trimethylsilyl)- | 27.498 | 73.05 |
| D-Turanose, heptakis(trimethylsilyl)- | 34.608 | 73.05 |
| D-Xylopyranose, 1,2,3,4-tetrakis-O-(trimethylsilyl)- | 28.452 | 73.1 |
| Glucose oxime hexakis(trimethylsilyl) | 20.544 | 73.1 |
| Inosose, 2-desoxy-, O-methyloxime, tetrakis-O-(trimethylsilyl)- | 21.802 | 73.1 |
| Maltose, octakis(trimethylsilyl)- | 33.223 | 204.1 |
| Melibiose, octakis(trimethylsilyl)- | 29.563 | 204.15 |
| .alpha.-D-Galactopyranoside, methyl 2,3,4,6-tetrakis-O-(trimethylsilyl)- | 30.694 | 204.1 |
| .alpha.-D-Galactoside, methyl tetrakis-O-(trimethylsilyl)- | 22.127 | 73.1 |
| .alpha.-D-Glucopyranoside, 1,3,4,6-tetrakis-O-(trimethylsilyl)-.beta.-D-fructofuranosyl 2,3,4,6-tetrakis-O-(trimethylsilyl)- | 33.483 | 361.2 |
| .alpha.-D-Glucopyranoside, methyl 2,3,4,6-tetrakis-O-(trimethylsilyl)- | 30.423 | 204.1 |
| .alpha.-D-Mannopyranoside, methyl 2,3,4,6-tetrakis-O-(trimethylsilyl)- | 41.75 | 204.15 |
| 2-O-Glycerol-.alpha.-d-galactopyranoside, hexa-TMS | 28.989 | 204.1 |
| Glycoside, alpha.-methyl-trtrakis-O-(trimethylsilyl)- | 20.798 | 73.1 |
| Hydroquinone-.beta.-d-galactopyranoside.pentakis(trimethylsilyl)- | 35.26 | 73.05 |
| Sugar alcohols | | |
| Arabinitol, pentakis-O-(trimethylsilyl)- | 16.326 | 73.1 |
| D-Glucitol, 6-deoxy-1,2,3,4,5-pentakis-O-(trimethylsilyl)- | 16.827 | 73.1 |
| D-Mannitol, 1,2,3,4,5,6-hexakis-O-(trimethylsilyl)- | 20.926 | 73.1 |
| D-Myo-Inositol, 1,2,4,5,6-pentakis-O-(trimethylsilyl)-, bis(trimethylsilyl) phosphate | 30.911 | 73.05 |
| Inositol, 1,2,3,4,5,6-hexakis-O-(trimethylsilyl)-, scyllo- | 23.714 | 73.1 |
| Myo-inositol, 5-deoxy-1,2,3,4,6-pentakis-O-(trimethylsilyl)- | 18.877 | 73.1 |
| Per-O-trimethylsilyl-(3-O-.beta.-d-mannopyranosyl-d-glucitol) | 42.387 | 73.1 |
| Ribitol, 1,2,3,4,5-pentakis-O-(trimethylsilyl)- | 21.302 | 73.1 |
| Other classes | | |
| (4-Hydroxy-3-methoxyphenyl)ethylene glycol tris(trimethylsilyl) ether | 39.646 | 297.2 |
| 1-Monooleoylglycerol trimethylsilyl ether | 38.308 | 129.1 |

| | | |
|--|--------|--------|
| 2(3H)-Furanone, dihydro-3,4-bis[(trimethylsilyl)oxy]-, cis- | 14.931 | 73.1 |
| 2(3H)-Furanone, dihydro-3,4-bis[(trimethylsilyl)oxy]-, trans- | 12.192 | 73.1 |
| 2,6,10,14,18,22-Tetracosahexaene, 2,6,10,15,19,23-hexamethyl-, (all-E)- | 40.733 | 69.1 |
| Butanal, 2,3,4-tris[(trimethylsilyl)oxy]-, O-methyloxime, [R-(R@,R@)]- | 23.454 | 73.1 |
| Butanal, 2,3,4-tris[(trimethylsilyl)oxy]-3-[[trimethylsilyl]oxy]methyl-, O-methyloxime, (S)- | 16.492 | 73.1 |
| Butane, 1,2,3,4-tetrakis[(trimethylsilyl)oxy]- | 11.437 | 73.1 |
| Octacosanol trimethylsilyl ether | 42.772 | 467.55 |
| Pseudo uridine penta-TMS | 14.013 | 73.1 |
| Salbutamol, N-trifluoroacetyl-o,o,o-tris(trimethylsilyl)deriv. | 42.965 | 369.25 |
| Silane, [(3,7,11,15-tetramethyl-2-hexadecenyl)oxy]trimethyl- | 27.816 | 143.15 |
| Silane, [[2-[3,4-bis[(trimethylsilyl)oxy]phenyl]-3,4-dihydro-2H-1-benzopyran-3,5,7-triyl]tris(oxy)]tris[trimethyl]-, (2R-cis)- | 41.053 | 73.1 |
| Trimethylsilyl ether of glycerol | 27.277 | 73.05 |
| Uridine, 2',3',5'-tris-O-(trimethylsilyl)- | 19.408 | 73.1 |

Conclusion

GC-MS based metabolite profiling approach was used to evaluate the metabolic content of Mango Cv. Kesar. The results from our present study demonstrated the presence of 85 metabolites including organic acids, Sugars, Sugar alcohols and others. The identified potential metabolites presence in leaf extract may be responsible to exhibit different biological significance in the agriculture, food industry, pharma industry etc. Further, isolation and biological evaluation of the potential significant compounds are explored for the discovery of drugs as well as to establish the traditional use of the plants.

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