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Abir EM Ajab Department of Chemistry, Kordofan University, Sudan

Mohammed E Osman

Department of Chemistry, College of Science, Sudan University of Science and Technology, Khartoum P.O. Box 407, Sudan

Elfatih A Hassan

Department of Chemistry, College of Science, Sudan University of Science and Technology, Khartoum P.O. Box 407, Sudan

Seifeldawla A Elhag Department of Chemistry, Sinnar University, Sudan

Correspondence Abir EM Ajab Department of Chemistry, Kordofan University, Sudan

Functional properties of *Acacia nilotica* var. *adstringens* gum

Abir EM Ajab, Mohammed E Osman, Elfatih A Hassan and Seifeldawla A Elhag

Abstract

This paper examines the emulsion stability and rheological properties of composite sample of *A. nilotica* var. *adstringens* gum from Kordofan state.

Emulsification study of *A. nilotica* var. *adstringens* gum showed that the gum has an excellent emulsifying stability. Rheological behavior of aqueous solutions of *Acacia nioltica* var. *adstringens* gum shows that the gum solution at 25% concentration indicates a shear-thinning at lower shear rates, and a viscosity plateau (Newtonian behavior) at higher shear rates. At 50% concentration the gum solutions exhibits Newtonian flow. The dynamic rheological behavior of gum solution at 50% concentration shows that storage modulus (G') was far less than loss modulus (G'') at all frequencies, energy of deformation is dissipated viscously and the sample exhibits liquid-like behavior.

Keywords: functional, properties, Acacia nilotica, kordofan

1. Introduction

Acacia adstringens is a subspecies of *A. nilotica* which is occasionally used for firewood and good quality charcoal in the Sahelian regions. It has a hard heavy heartwood, which resistants water and termites. The subspecies uses include construction work, boat-building, fencing, tool handles and art objects. It is a source of gum and tannin, the gum is locally used for making ink. The subspecies is commonly, planted as shade tree.

Many food products, in the markets, are in the emulsion state such as cheese, milk, salad dressings, sauces, beverages and coconut milk (Gonzalez 1991)^[4]. An emulsion is a dispersed system that consists of two immiscible liquids (usually oil and water), with one of the liquids dispersed as small droplets in the other called continuous phase. Gums are, extensively, used as an emulsifier/stabilizer in beverage emulsion for soft drinks (Tan, 2004)^[13].

The Arabinoglactan protein (AGP) fraction has been shown to be responsible for the emulsifying properties of gum arabic (Randall *et al.*, 1988; Randall *et al.*, 1989)^[8, 9].

The function of emulsifier or stabilizer is essentially to increase the viscosity of the aqueous phase by thickening it so that it approximates or slightly exceeds that of the oil.

During the process of making a mixture of two immiscible substances, minimal contact between the oil phase and water can, initially, be achieved by the formation of small, spherical, droplets through the input of work and addition of emulsifier, facilitating the formation of these small particles by reducing interfacial tension. In fact, the aim of emulsification such as O/W is to produce a small size of droplets of the dispersed phase (oil) and the size of the droplets should not be increased during emulsion storage. This means that emulsions have to be stable for a certain period of time. Gums form a protective layer around the oil droplets that prevents droplets from aggregating, flocculating and coalescing. It also reduces the oil water interfacial tension, thereby facilitating the disruption of emulsion droplets during homogenization.

The gum from *Acacia senegal* has a functional ability to act as emulsifier that stabilizes oil-inwater emulsion (Yokoyama *et al* 1988, Randall *et al* 1988) ^[16, 7]. It is now known that the protein-rich high molecular mass component adsorbs preferentially onto the surface of the oil droplets. It is envisaged that the hydrophobic polypeptide chains adsorb and anchor the molecules to the surface while the carbohydrate blocks inhibit flocculation and coalescence through electrostatic and steric repulsions (NGARA, 2005). Therefore, the protein moiety affects the emulsifying behavior of gums and result in emulsion capacity and stability which is exhibition gums with high nitrogen content (Randall *et al* 1988, Dickenson 1992) ^[7, 9]. The emulsifying properties of gums, however, are, directly, (NGARA, 2005) ^[6] influenced by the botanical type, the nature of soils and the climate. In addition to the emulsifying properties of *Acacia* gums, other factors related to the emulsification process influence the properties of an emulsion (Brosel and Schubert, 1999) ^[1].

Rheology is the science of deformation and flow. Rheological measurements have also been considered as an analytical tool to provide fundamental insight on the structural organization of food. The degree of thickening depends on the type of gum and the level of concentration. It has been reported that gum arabic solutions show Newtonian flow behaviour in the shear rate range from 50s⁻¹ to 100s⁻¹, at gum concentration as low as 4%. (Mothé et al., 1999) ^[5]. In the shear rate range from 1 s⁻¹ to 50s⁻¹, the shear thinning behavior has been observed. In the gum concentration range 10 to 25%, gum solutions show a significant shear thinning behavior, as compared with that of solutions with concentration in the range 30-50 %. (Mothé et al., 1999)^[5]. Time dependent thickening flow behavior has been observed for 3 wt% gum arabic solutions at shear rate below 1 s⁻¹. Gum arabic solutions have shown shear thinning behavior at concentrations between 3 and 32 wt%. (Sanchez et al., 2002) [10]. Gum solution above 30% concentration shows high solution viscosity and exhibits pseudo plasticity. (Williams et al., 1990)^[15].

The rheological complexity of gum arabic is also evident from the increase of viscosity/elasticity with time, which has been attributed to the interfacial activity of gum arabic, gradually developing an elastic interfacial film between samples and measuring geometry. (Sanchez *et al.*, 2002)^[10].

2. Materials and Methods

2.1 Materials

One composite sample of *Acacia nilotica* var. *adstringens* gum collected from North Kordofan state, Sudan, seasons 2016 and 2017 was used.

2.2 Methods

2.2.1 Emulsion preparation

Distilled water was added to 12 g of the gum sample (based on dry weight) to become 40 g in total with a concentration of 30 % (w/w) gum solution. The sample was agitated on a tube roller mixer overnight until the sample completely dissolves. Exact calculated grams for each samples (in the range from about 27 to about 28 g) of the prepared gum solution was filtered using 100 μ m mesh, then mixed with 0.52 ml of 10 % (W/V) sodium benzoate solution as a preservative, and 0.48 ml of 10 % (W/V) citric acid solution to adjust the pH to 4, distilled water was added until the total weight become 32.0 g. then, 8.0 g ODO oil was added to the gum solution to give a total of 40 g and final concentration of 20%.

The mixed solution was homogenized for 3 minutes using a polytron (PT-2100) homogenizer at 26000 rpm. Impeller (PTDA21 9 mm tip diameter) was used as dispersing tool. To achieve small particle size < 1 micron, the pre-emulsified mixture was homogenized using a high-pressure Nanomizer (NM2-L100, Yoshida a kikai Co. Ltd.). In order to achieve effective disaggregation of the gum, the mixture passed, twice, at 50 MPa.

The final emulsion was divided in two aliquots and kept in closed glass vials. One of the aliquots was kept at 60°C in the Vacuum Oven (Gallenkamp. OVA031. XX1.5), and the other one at room temperature.

2.2.1.1 Droplet size analysis

Droplet size distribution of the emulsions was analyzed using Mastersizer 3000, a laser diffraction particle size analyzer (Malvern Instruments). Distilled water was used as dispersant and a value of 1.450 was used for the refractive index for oil phase (ODO). Emulsification stability of samples kept at 60°C was evaluated by particle size change after accelerated stability test for 3 and 7 days.

A particle size emulsion is described by the volume median diameter (VMD).

2.2.2 Rheological measurments

Rheological measurements were carried out using KINEXUS Pro⁺ (Malvern Instruments) fitted with cone and plate geometry with a cone diameter of 40 mm and an angle of 2^0 . Steady shear viscosity curves were measured for gum solutions 25 and 50 % w/v both upon shear rate ramp-up (from 0.01 to 10000 s⁻¹) and subsequent shear rate ramp-down (from 10000 back to 0.01 s⁻¹). Dynamic rheological measurements, to determine the elastic modulus (G'), viscous modulus (G'') and dynamic viscosity, were performed in the frequency range of 0.1–10 Hz. The linear viscoelastic region was assessed, at 1 Hz. The temperature of the samples were controlled within 0.1°C using a Peltier element. The rheometer control and data processing was done by computer software (Rheology Advantage Data Analysis Program, TA).

3. Results and Discussion

3.1 Emulsification of A. nilotica var. adstringens

Table 1 and Fig 1 show the span% and mean diameters of cumulative droplet distributions (d0.1, d0.5 and d0.9) measurement of *Acacia nilotica* var. *adstringens* emulsions at different conditions (fresh emulsion and after storage for 3 and 7 days at 60° C respectively).

The fresh emulsion and emulsion stored for 7 days at 60° C show low degree of polydispersity (span %), indicating good uniformity of the droplet size. The span change varies from 2.537 in fresh emulsion to 2.829 for emulsion stored for 7 days at 60° C. However, the specific surface area decreased when the emulsion stored for 7 days at 60° C

Fig 1 shows the particle size distribution of *A. nilotica* var. *adstringens* emulsion (freshly prepared emulsion, emulsion stored for 3 and 7 days at 60° C) is constant (~ 1 micron), indicating a stable emulsion. The particle size of the emulsion was described as volume median diameter (VMD), for *A. nilotica* var. *adstringens* (VMD) was found 0.532 μ m for fresh emulsion and 0.557 μ m for emulsion stored for 7 days at 60° C. It is generally known that an increase in droplet mean diameter upon storage is an indicator of droplet coalescence. In relation to emulsion rheology, the occurrence of droplet coalescence will be normally accompanied by a decrease in the emulsion viscosity.

Table 1: Span% and mean diameters of cumulative droplet distributions (μm) of freshly and stored (3 and 7 days at 60° C) emulsions of *Acacia nilotica* var. *adstringens* (whole gum)

	Freshly prepared	3 days at 60°C	7 days at 60°C
Span %	2.537	416.2	2.829
D [32]	0.0781	0.177	0.0784
D [43]	0.588	146	0.771
d (0.1)	0.0249	0.0487	0.0247
d (0.5)	0.532	1.12	0.557
d (0.9)	1.37	465	1.6

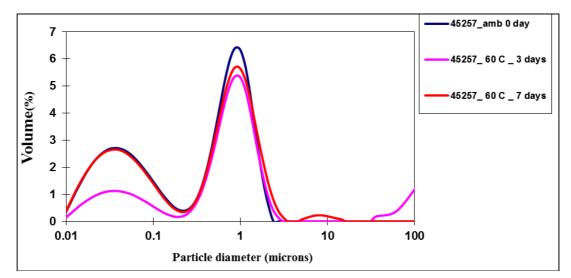


Fig 1: Particle size distributions of Acacia nilotica var. adstringens emulsion

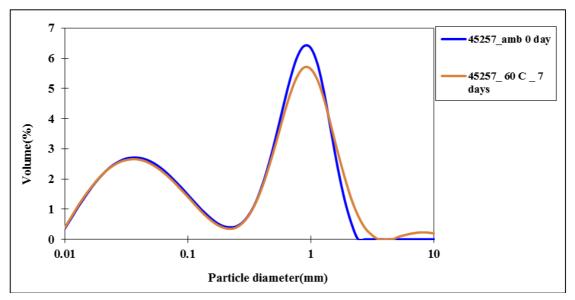


Fig 2: Particle size distributions of Acacia nilotica var. adstringens emulsion samples (freshly and 7days at 60° C)

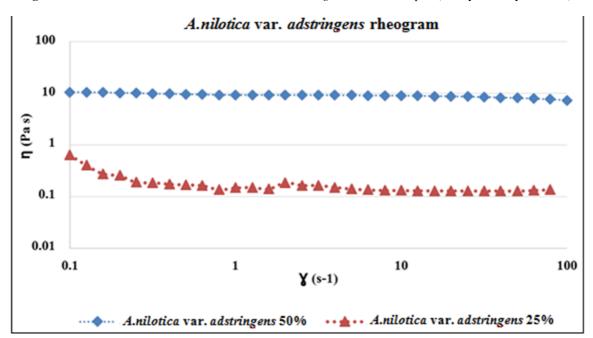


Fig 3: Viscosity-Shear rate profile of Acacia nilotica var. adstringens gum solution at 25% and 50% concentration

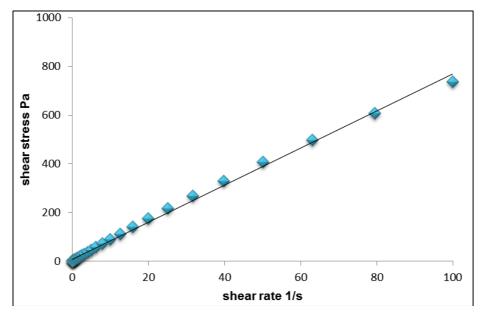


Fig 4: The shear rate vs. shear stress of A. nilotica var. adstringens gum 50%

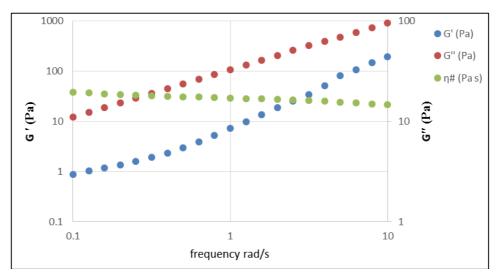


Fig 5: The dynamic rheological behaviour of Acacia nilotica var. adstringens gum solution at 50% concentration.

3.1.1 Emulsion stability index of *Acacia nilotica* var. *adstringns*

Acacia nilotica var. adstringens gum has a high molecular weight of AGP. The stability of emulsion offered by gum of *A. nilotica* var. adstringens may be due to the AGP fraction increase. It established that there is a relationship between the proportion of AGP component and emulsion stability (Dickinson *et al*, 1988) ^[3].

Fig 2 shows particle size distributions of *Acacia nilotica* var. *adstringens* emulsion samples (freshly and 7days at 60° C) which clearly showed that no change was found in most emulsions during the incubation for 7 days at 60° C.

3.2 Rheology

Figures 3 show the effect of shear rate on viscosity, at 25°C, for *Acacia nilotica* var. *adstringens* gum solutions of different concentrations. The flow curves at increasing shear rates showed that the apparent viscosity (η_0 , Pa.s) decreased slightly as the shear rate (γ s⁻¹) increased. A Newtonian low-shear plateau (the so-called zero-shear viscosity) was observed even at shear rates as low as 0.1s⁻¹. At intermediate shear rates (above roughly 1 s⁻¹) a trend to a high-shear Newtonian plateau was observed. The flow curve show a shear-thinning at lower shear rates, and a viscosity plateau

(Newtonian behavior) at higher shear rates (Fig 3). Similar rheological observations have been already reported for gum arabic (Williams et al 1990, Mothé et al., 1999 and Sanchez et al., 2002) ^[15, 5, 10]. The extent of the shear thinning seems to be reduced at higher concentrations. That may be regarded as from modifications in the macromolecular arising organization of the solution as the shear rate changes. At concentrations of 50% solutions, gum exhibit Newtonian flow. Therefore, the rheological behavior of gum solution approach Newtonian character as the gum concentration increases, Figure 3. Newtonian behaviors were expected as a result of the relatively compact conformation of branched gum (Tanaka et al., 2006) ^[14]. This result is similar to that obtained by (Satti, 2011) ^[11] for A. nilotica var. nilotica gum at 50% concentration.

Fig 4 shows that *A. nilotica* var. *adstringens* gum has a typical Newtonian behavior.

3.2.1 Dynamic rheological behavior

Fig 5 shows the dynamic rheological behavior of *A. nilotica* var. *adstringens* gum solution at 50wt% concentration. The loss modulus (G") was higher than storage modulus (G') at all frequencies. The energy used to deform the material is dissipated viscously and the sample exhibits liquid-like

behavior. The moduli showed less frequency dependence at lower frequency range and relatively higher frequency dependence at higher frequency range.

4. Conclusion

- The study shows that *Acacia nilotica* var. *adstringens* gum has a functional ability to act as emulsifier and stabilizes oil in water emulsion similar to *A. senegal*.
- Gum of acacia *adstringens* exhibit Newtonian flow at high concentration.
- The loss modulus of *Acacia adstringens* gum is higher than the storage modulus, which indicates liquid like behavior.

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6. References

- 1. Brosel S, H Schubert. Investigations on the role of surfactants in mechanical emulsification using a high-pressure homogenizer with an orifice valve, Chem. Eng. Proc. 1999; 38:533-540.
- 2. Dickinson E. Hydrocolloids at interfaces and the influence on the properties of dispersed system, Food Hydrocolloids. 2003; 17:25-39.
- 3. Dickinson E, Murray BC, Stainsby G, Anderson DMW. Food Hydrocolloids. 1988; 3:65.
- 4. Gonzalez ON. Coconut milk: Excerpts from Coconut as Food. Phil. J. Coco. Stud. 1991; 16:47-55.
- 5. Mothé CG, Rao MA. Food Hydrocolloids. 1999; 13:501.
- 6. NGARA. Network for Natural Gums and Resins in Africa, Publication Series 3, Kenya Forestry Research Institute, Nairobi, Kenya, 2005.
- 7. Randall RC, Phillips GO, Williams PA. The role of the proteinaceous component on the emulsifying properties of gum Arabic, Food Hydrocolloids. 1988; 2(2):131-140.
- 8. Randall RC, Phillips GO, Williams PA. Fractionation and characterization of gum from *Acacia Senegal*, Food Hydrocolloids. 1989; 3(1):65-76.
- 9. Randall RC, GO Phillips, PA Williams. The role of the proteinaceous component on the emulsifying properties of gum Arabic, Food hydrocolloids. 1988; 2:131-140.
- 10. Sanchez C, Renard D, Robert P, Schmitt C, Lefebvre J. Food Hydrocolloids. 2002; 16:257.
- 11. Satti AAE. Characterisation and toxicological study of *Acacia nilotica* var. *nilotica* gum from Sudan, PhD thesis, Faculty of Science, chemistry department, Sudan University of Science and Technology, 2011.
- 12. Tadros T. Application of rheology for assessment and prediction of the long-term physical stability of emulsions, Advances in Colloids and Interface Science. 2004; 108–109:227–258.
- 13. Tan CT. Beverage emulsions. In S. E. Friberg, K. Larsson, &J. Sjoblom (Eds.), Food emulsions. Boca Ration: CRC Press, 2004, 485–524.
- 14. Tanaka S, Fang Y, Nishinari K. Foods and Food Ingredients. 2006; 211:216.
- Williams GR. In Pillips, G. O., Wedlock, D. J., Williams, P. A., (eds) " Gums and stabilizers for the Food Industry 5.IRL Press, UK, 1990, 37.

 Yokoyama A, KP Srinivasan, HS Fogler. Stabilization mechanism of colloid suspension by gum Tragancanth: The influence of pH on stability. J Colloid Interface Sci. 1988; 126:141-149.